



## Solid Oxide Electrolysis Cells - High pressure operation

Ebbesen, Sune Dalgaard

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# **Solid Oxide Electrolysis Cells**

-High pressure operation

Sune D Ebbesen

Department of Energy Conversion and Storage

Danish Technical University, DTU

SYMPOSIUM

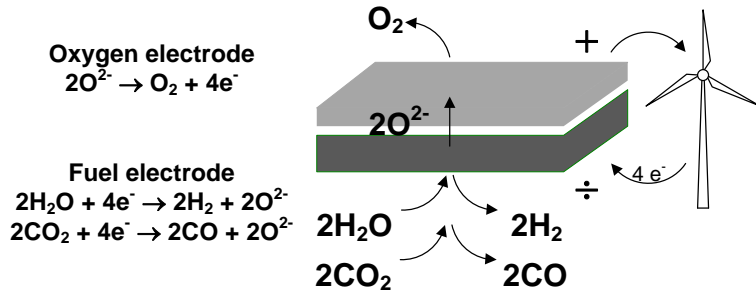
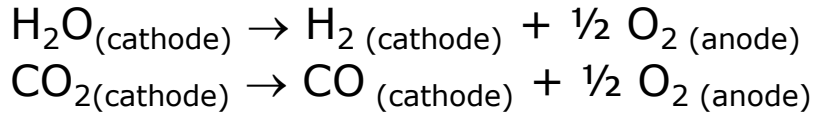
Water electrolysis and hydrogen as part of the future Renewable Energy  
System

Copenhagen, May 10-11 2012

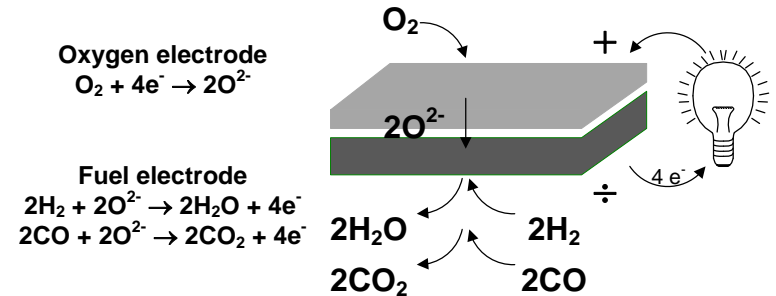
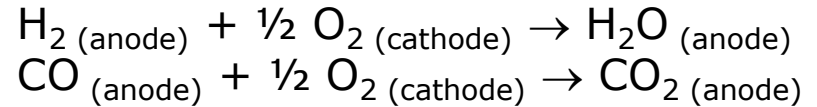
# The Solid Oxide Cell (SOC)

## — Reversible, SOEC $\leftrightarrow$ SOFC

- Electrolysis Cell (SOEC)



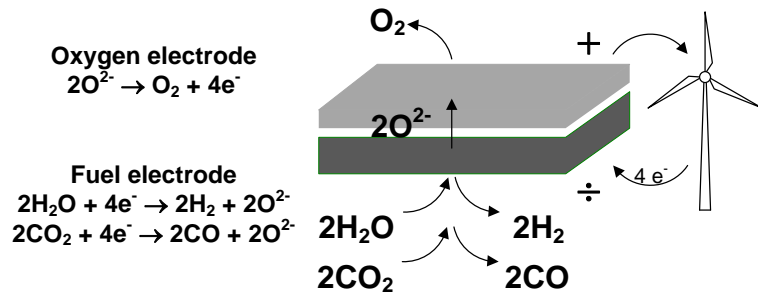
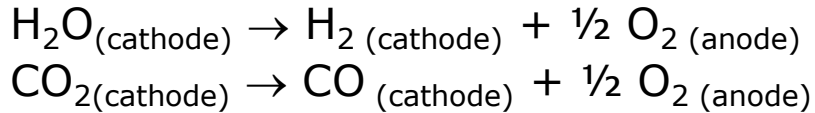
- Fuel Cell (SOFC)



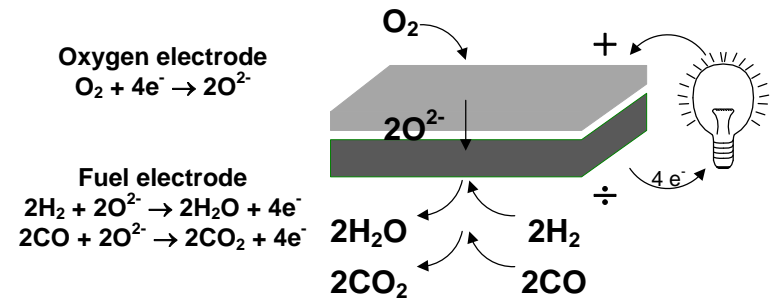
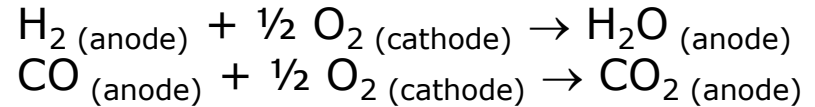
# The Solid Oxide Cell (SOC)

— Reversible, SOEC  $\leftrightarrow$  SOFC

- Electrolysis Cell (SOEC)



- Fuel Cell (SOFC)

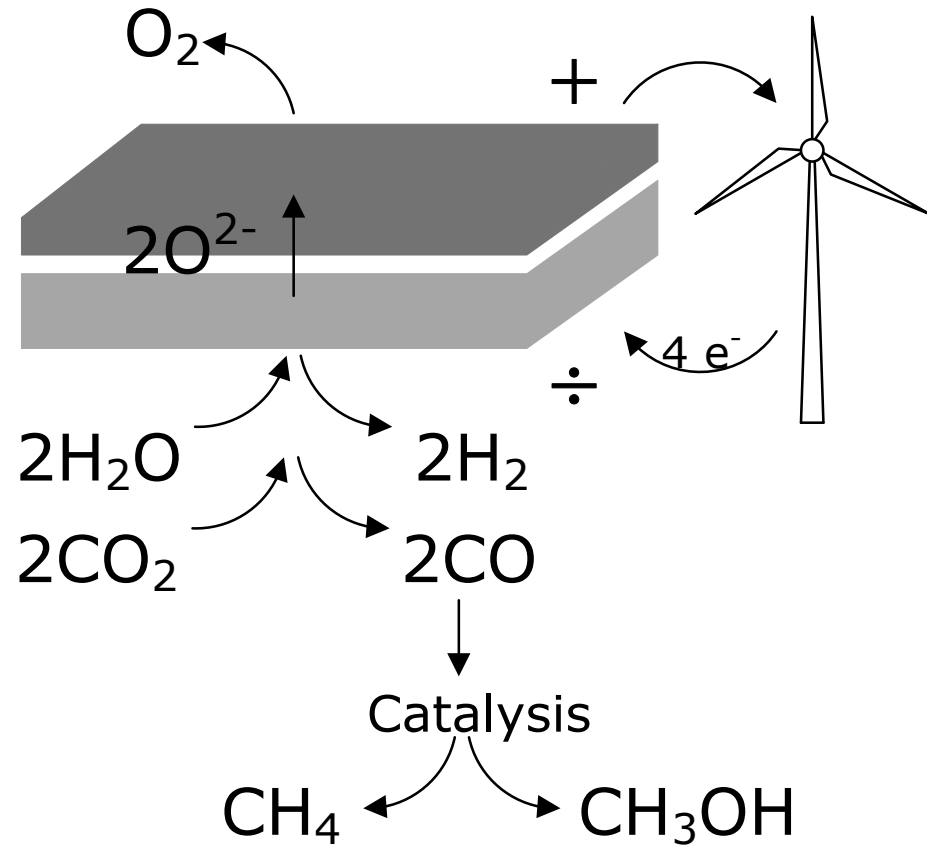


One major advantage of SOECs is the possibility to reduce  $\text{CO}_2$  to  $\text{CO}$

# Solid Oxide Electrolysis Cells

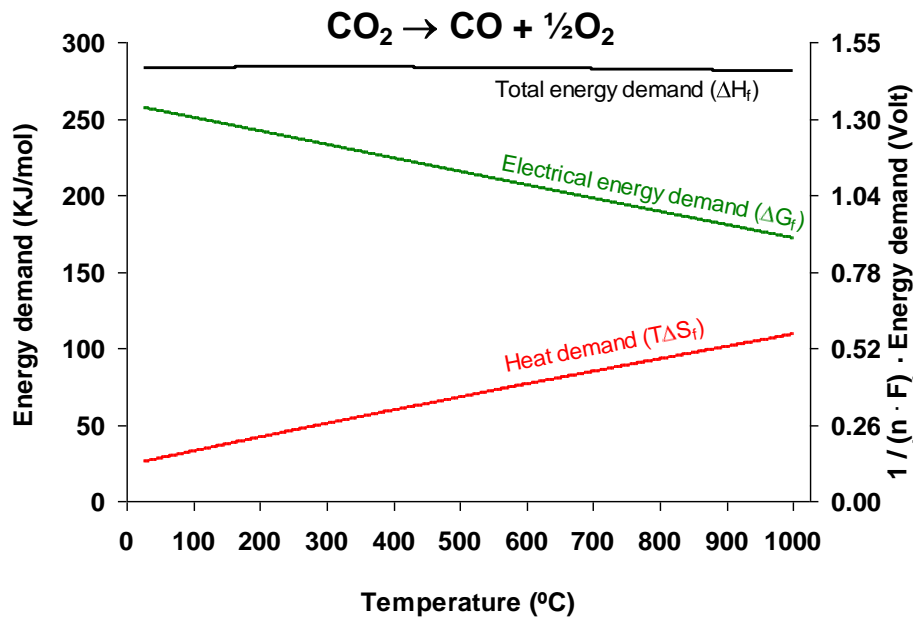
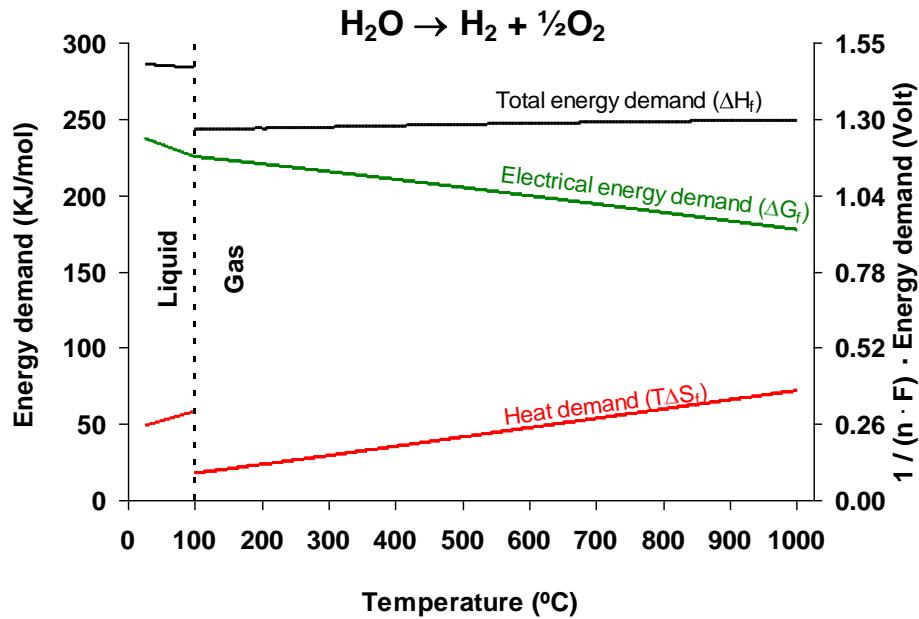
Oxygen electrode  
 $2\text{O}^{2-} \rightarrow \text{O}_2 + 4\text{e}^-$

Fuel electrode  
 $2\text{H}_2\text{O} + 4\text{e}^- \rightarrow 2\text{H}_2 + 2\text{O}^{2-}$   
 $2\text{CO}_2 + 4\text{e}^- \rightarrow 2\text{CO} + 2\text{O}^{2-}$

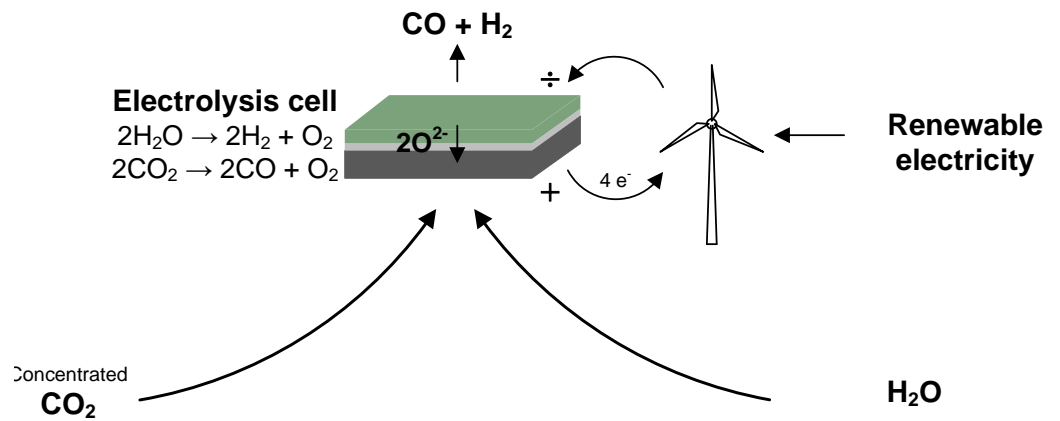


# Solid Oxide Electrolysis Cells

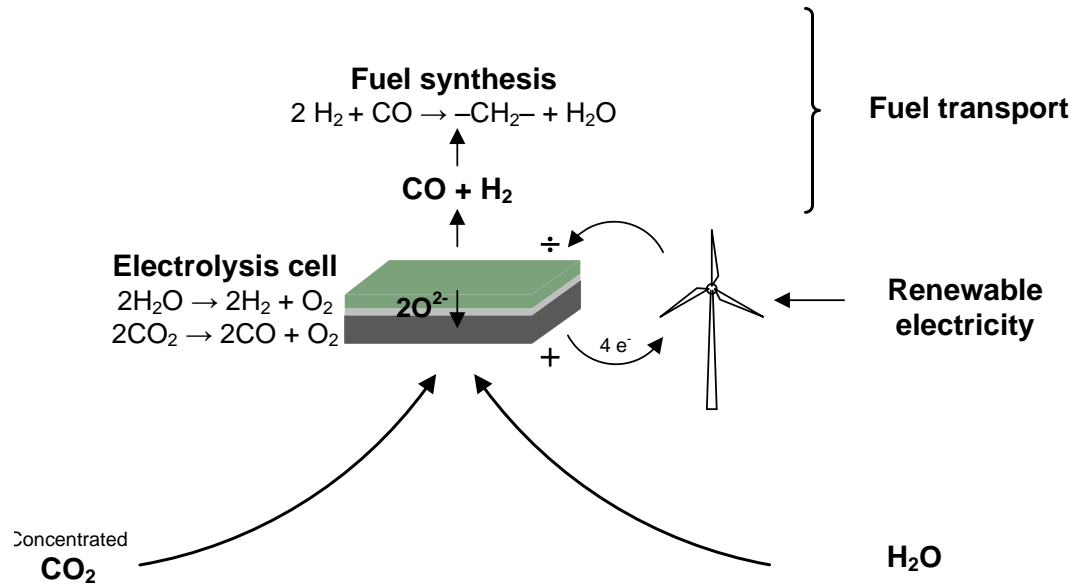
— Operation at high temperature



# Vision

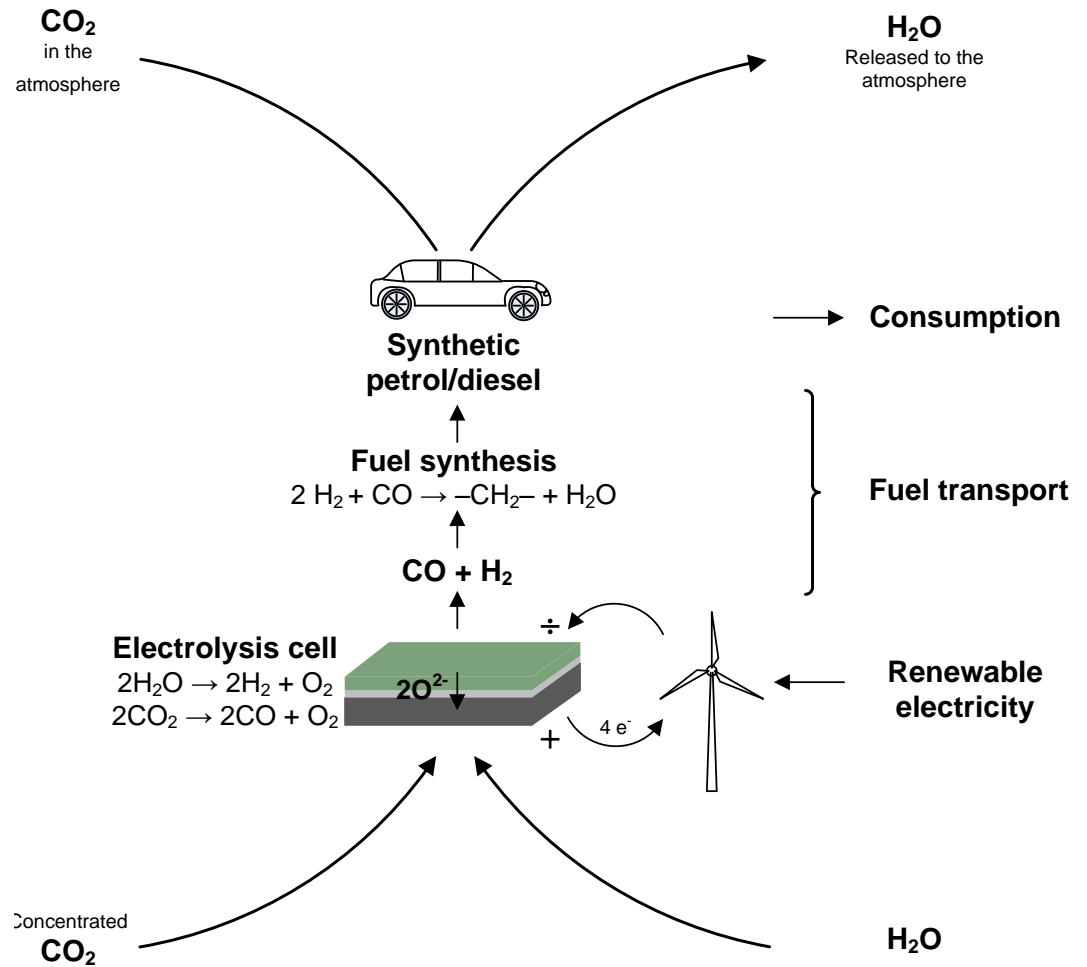


# Vision



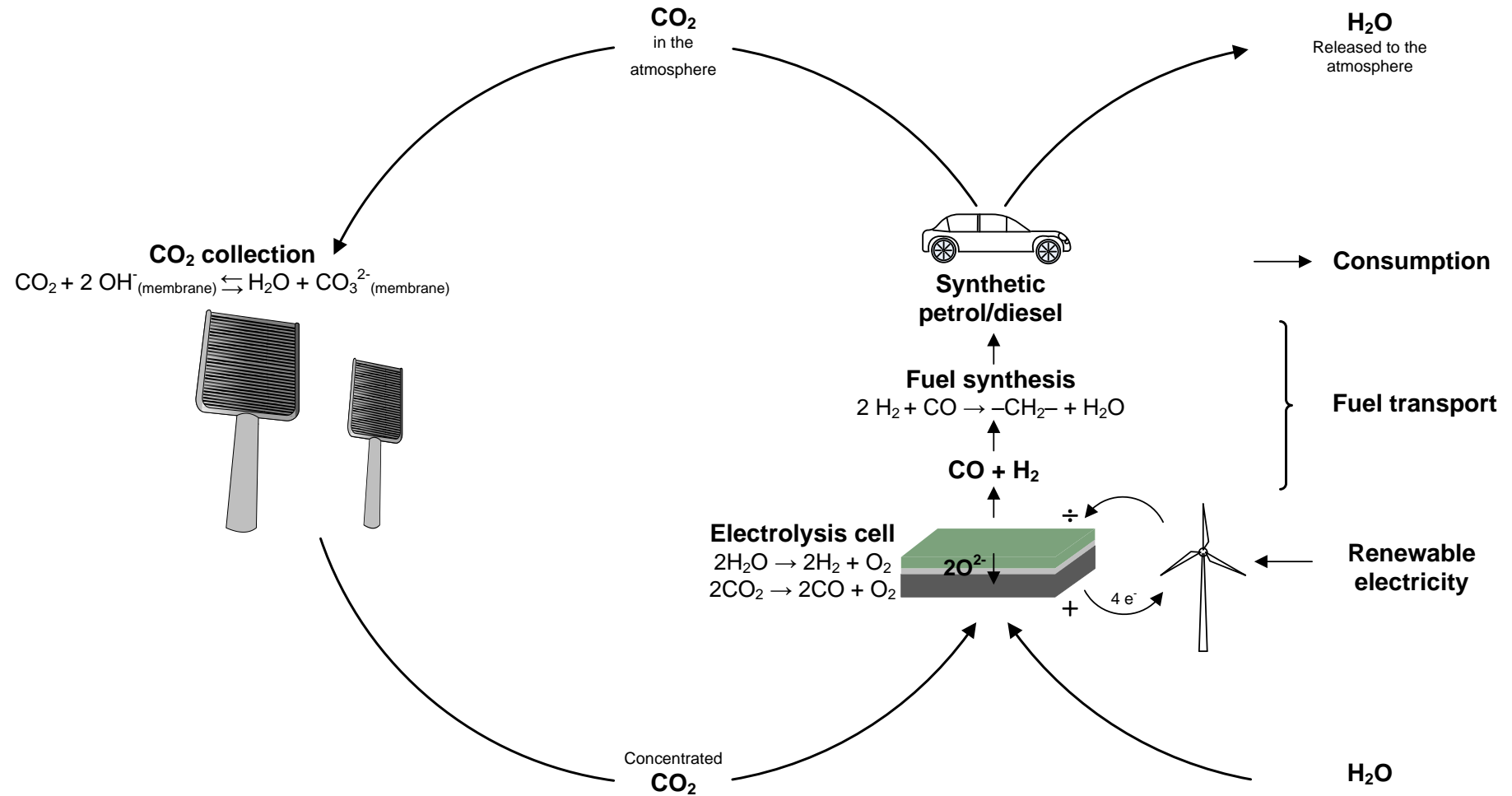


# Vision



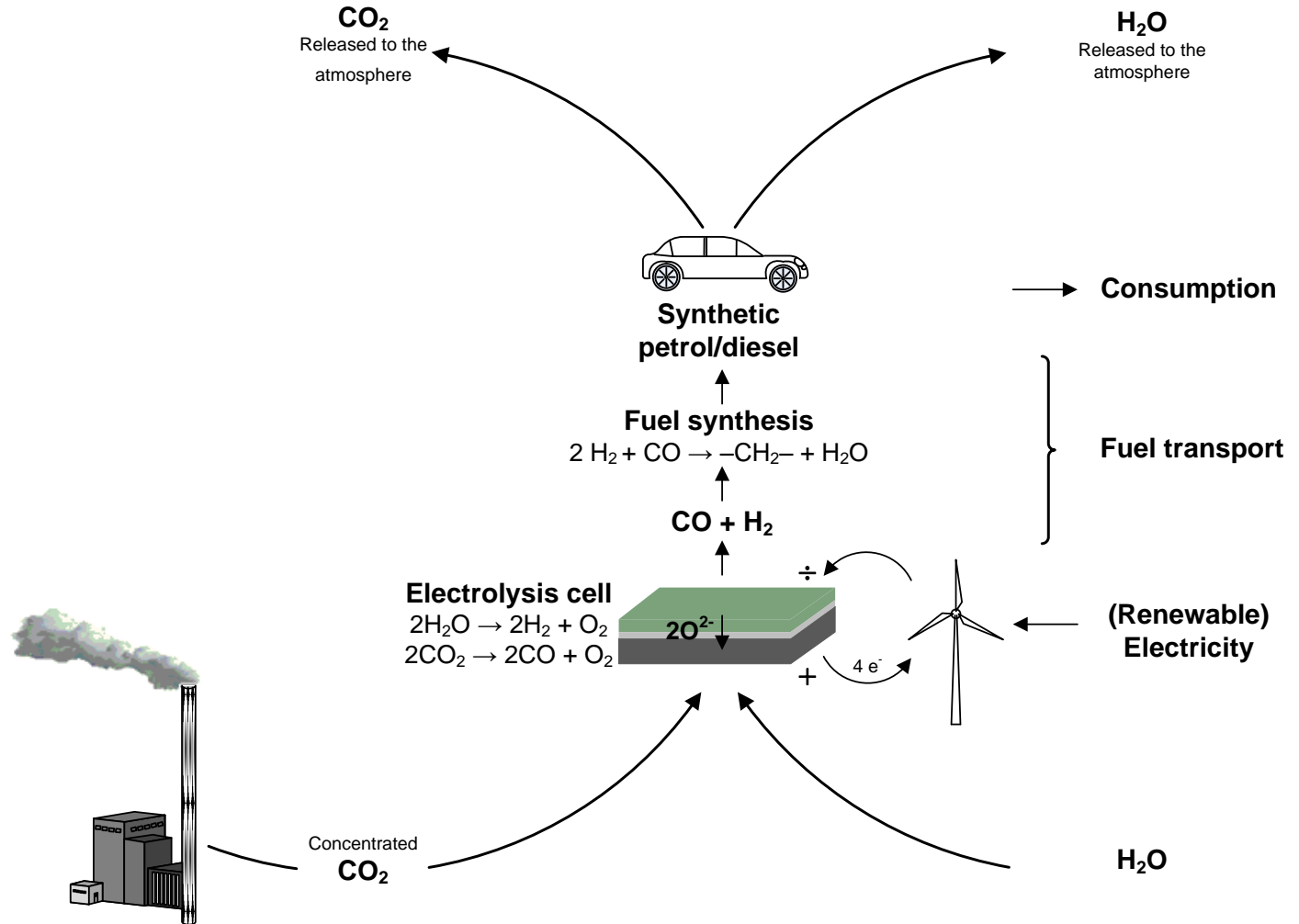
# Vision

## — Collection of CO<sub>2</sub> from the atmosphere



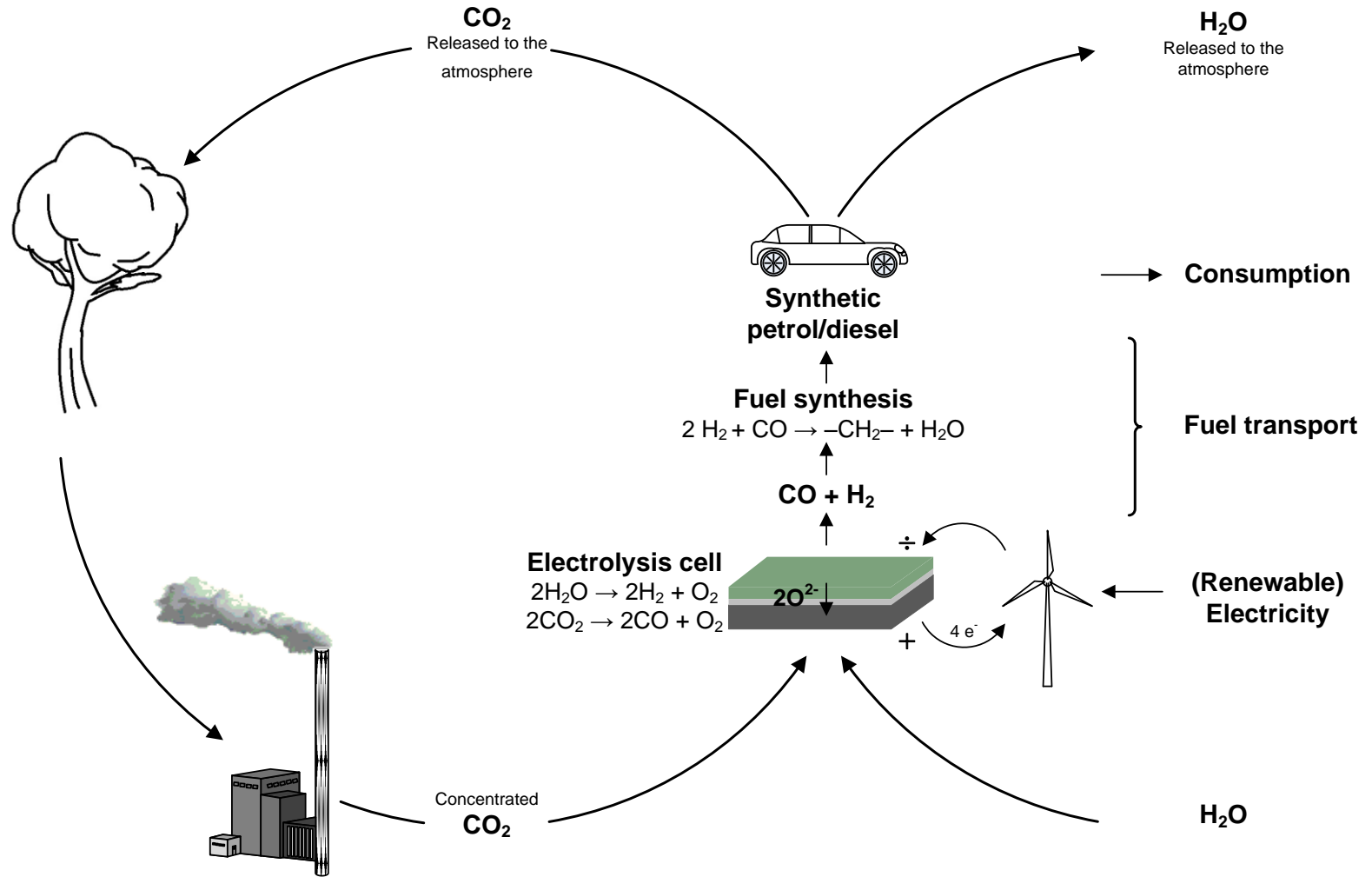
# Vision

## — Collection of CO<sub>2</sub> from industries



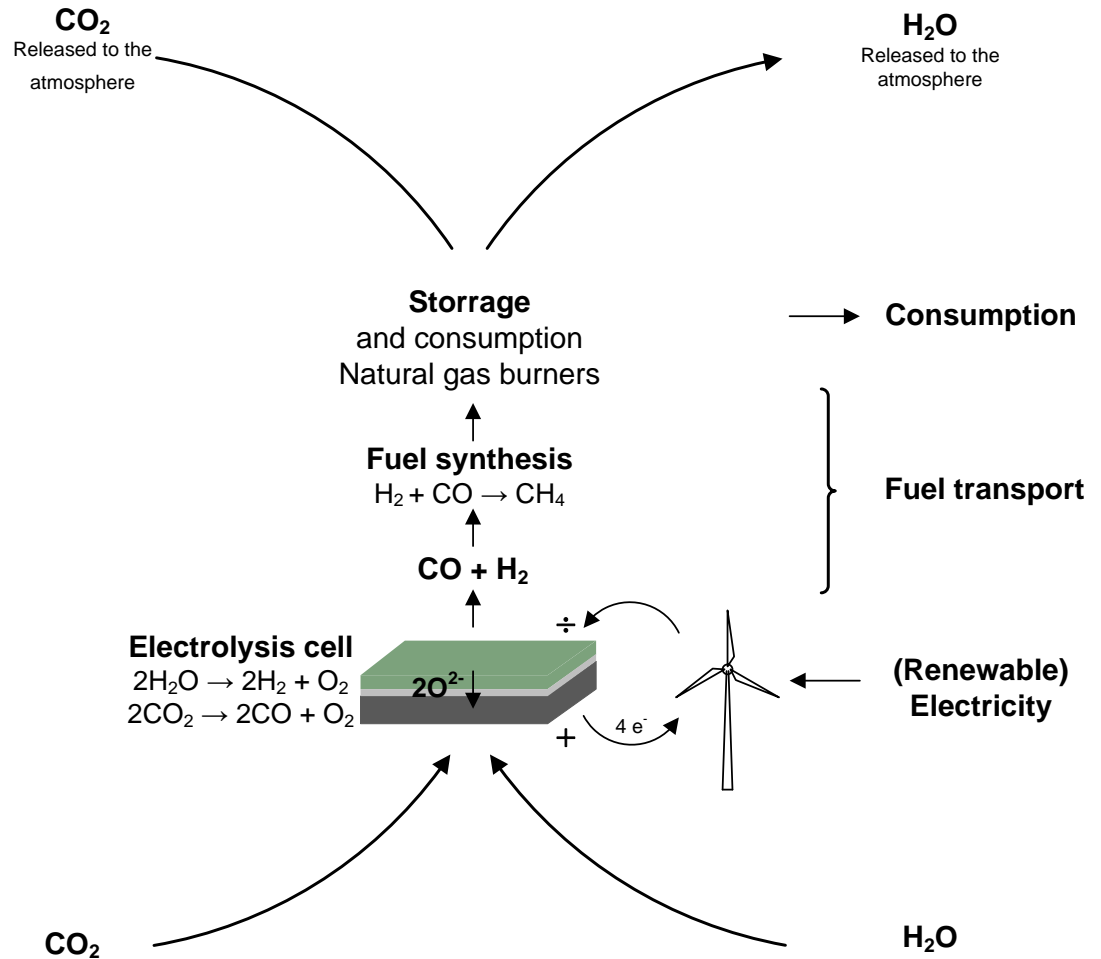
# Vision

## — Collection of CO<sub>2</sub> from power plants



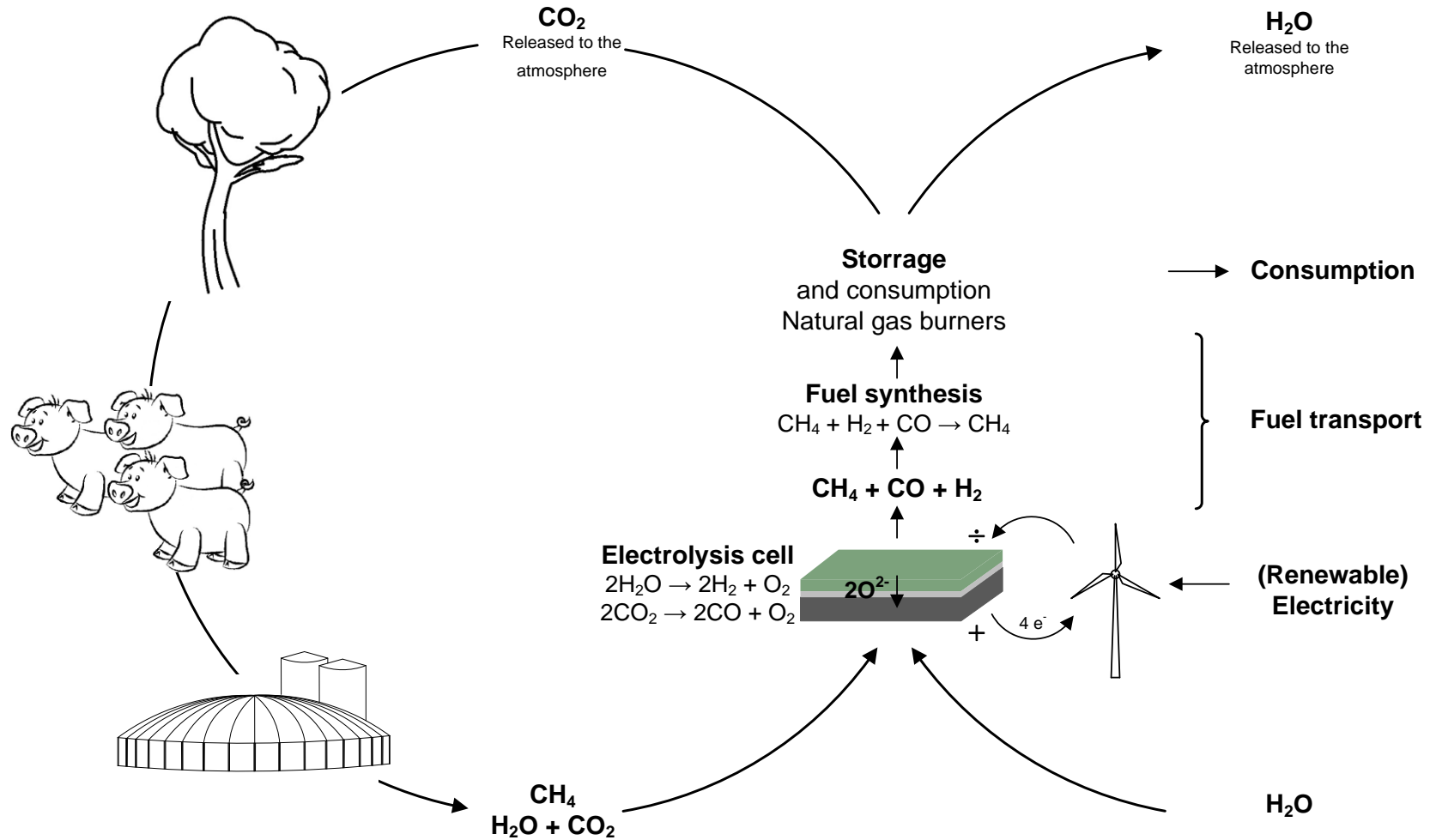
# Vision

## — Storing renewable electricity via Natural Gas



# Vision

## — Biogas upgrading



# Vision

- Production of synthetic fuels from renewable electricity (wind) and:
  - CO<sub>2</sub> from the atmosphere
  - CO<sub>2</sub> from the industry
  - CO<sub>2</sub> from biomass fired power plants
- Storage of renewable electricity via synthetic fuels and the natural gas grid
- Biogas upgrading

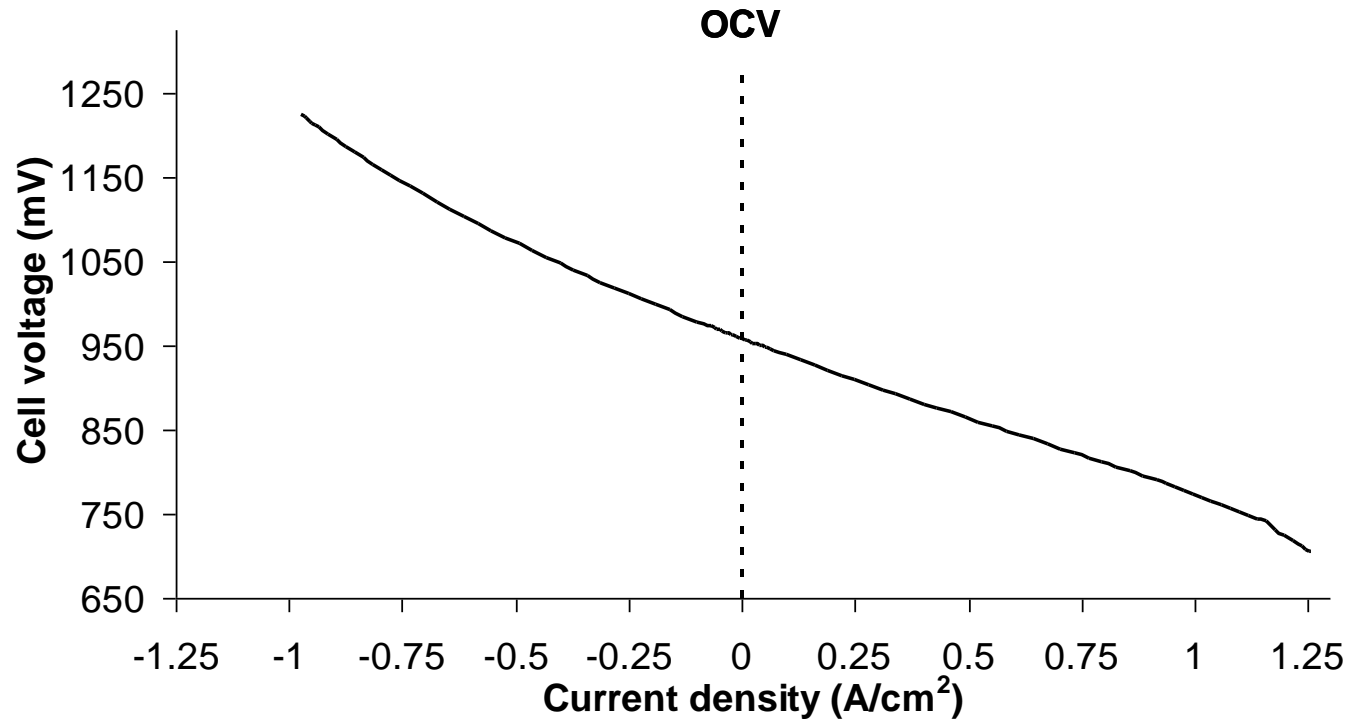
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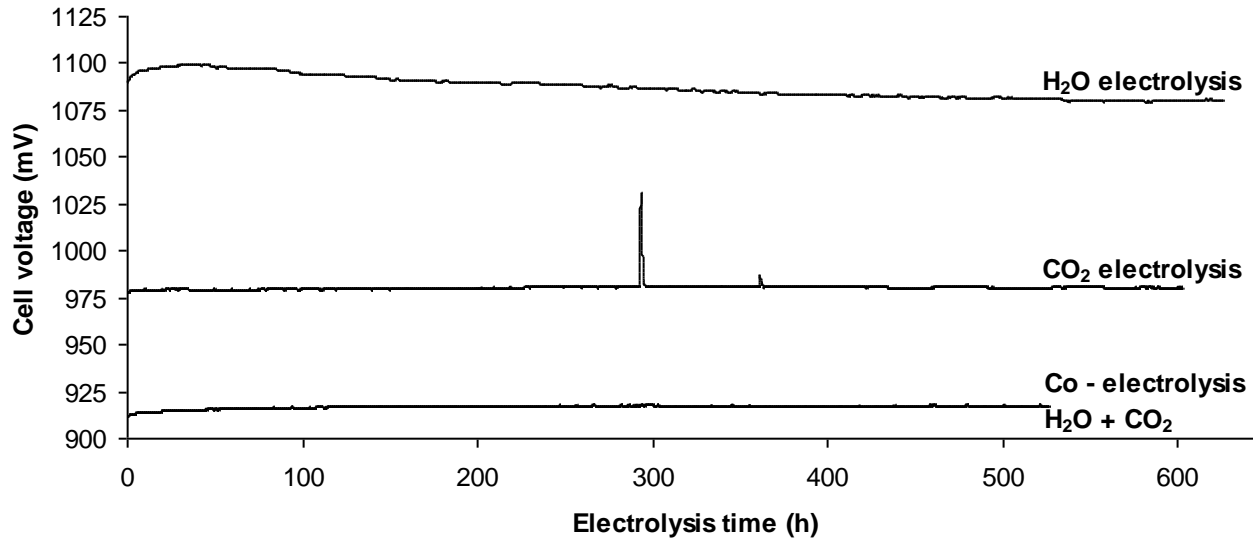


Conditions: 850°C, 50% H<sub>2</sub>O – 50% H<sub>2</sub>



# Electrolysis durability at low current density

—Cleaned inlet gases and improved setup



## Conditions:

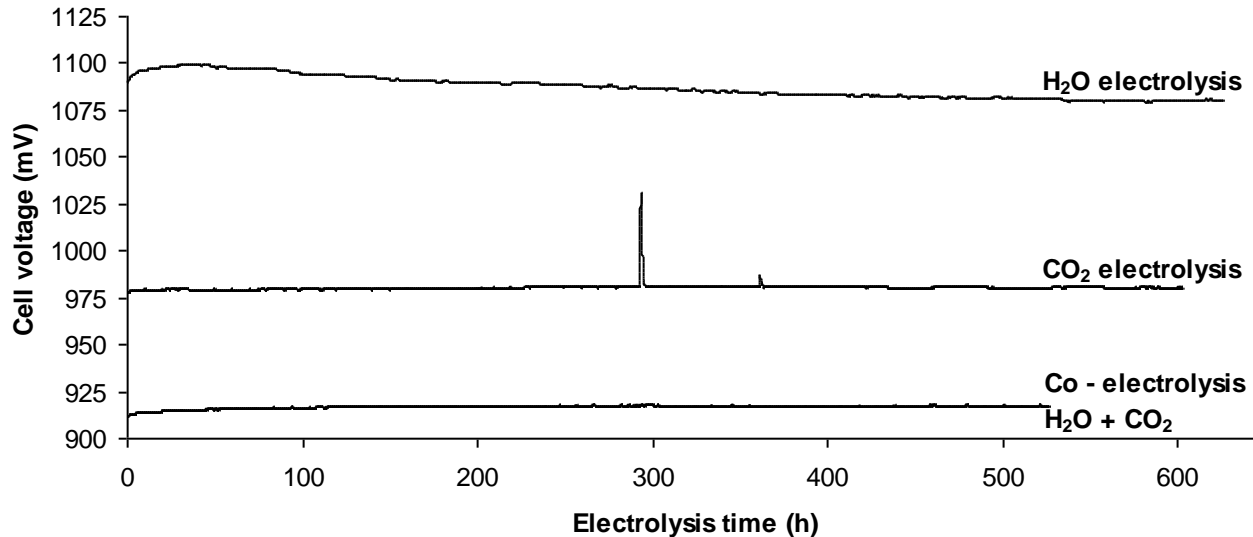
Steam electrolysis: 850°C, -0,50 A/cm<sup>2</sup>, 50% H<sub>2</sub>O – 50% H<sub>2</sub>

CO<sub>2</sub> electrolysis: 850°C, -0,25 A/cm<sup>2</sup>, 70% CO<sub>2</sub> – 30% CO

Co-electrolysis: 850°C, -0,25 A/cm<sup>2</sup>, 45% CO<sub>2</sub> – 45% H<sub>2</sub>O – 10% H<sub>2</sub>

# Electrolysis durability at low current density

—Cleaned inlet gases and improved setup



No degradation or even activation with clean inlet gases and a setup without contaminants

## Conditions:

Steam electrolysis: 850°C, -0,50 A/cm<sup>2</sup>, 50% H<sub>2</sub>O – 50% H<sub>2</sub>

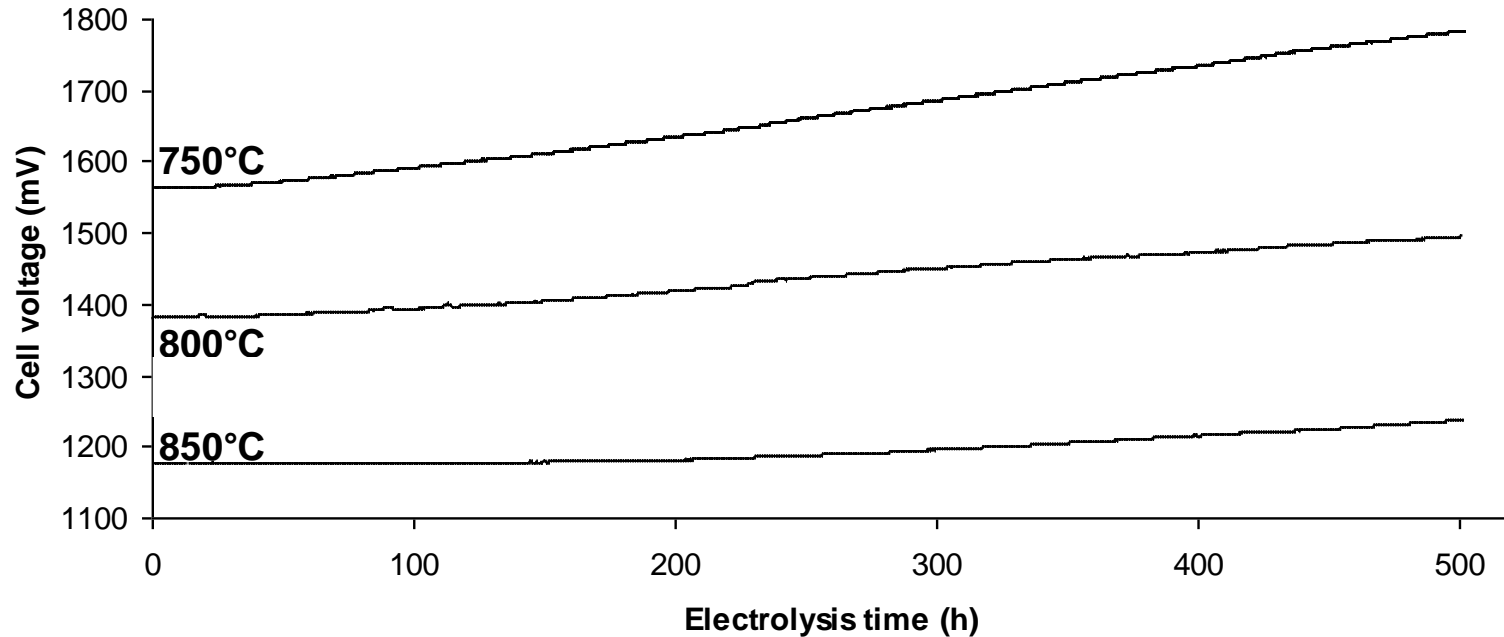
CO<sub>2</sub> electrolysis: 850°C, -0,25 A/cm<sup>2</sup>, 70% CO<sub>2</sub> – 30% CO

Co-electrolysis: 850°C, -0,25 A/cm<sup>2</sup>, 45% CO<sub>2</sub> – 45% H<sub>2</sub>O – 10% H<sub>2</sub>

# **Electrolysis durability at high current density**

# Electrolysis durability at high current density

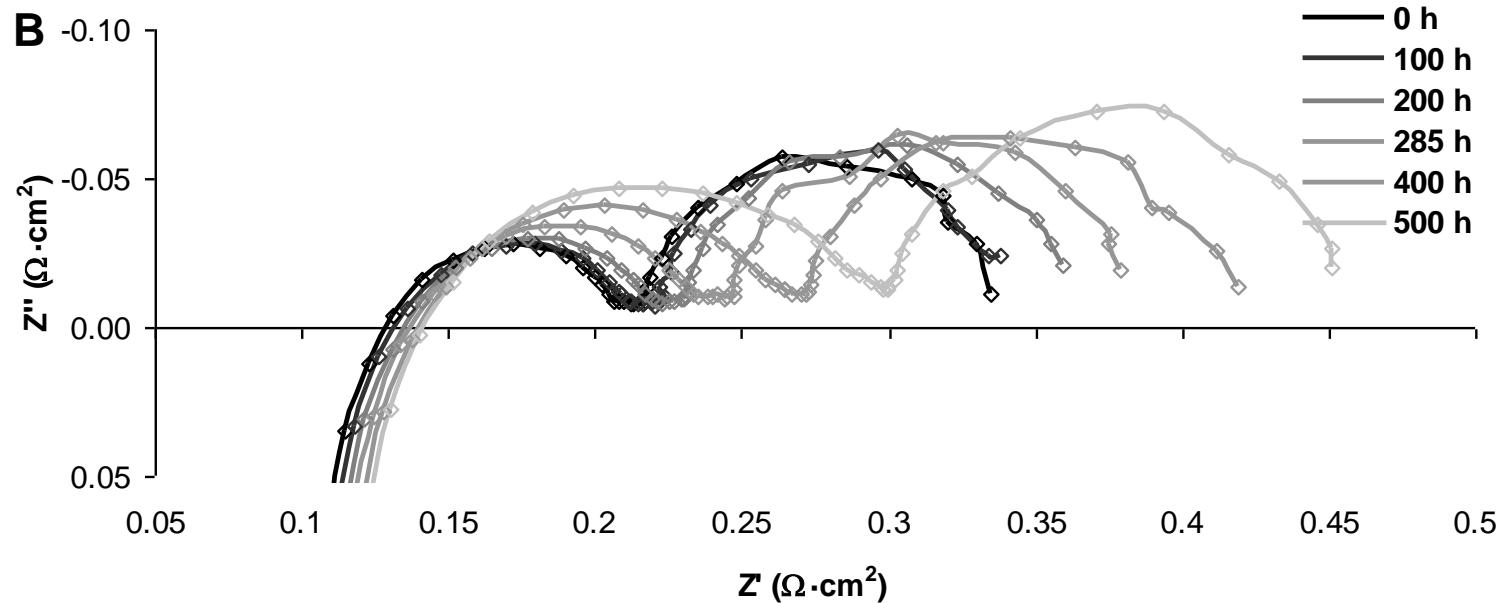
—Standard Ni-YSZ based cells with LSM O<sub>2</sub> electrode



Conditions: -1,0 A/cm<sup>2</sup>, 45% CO<sub>2</sub> – 45% H<sub>2</sub>O – 10% H<sub>2</sub>

# Electrolysis durability at high current density

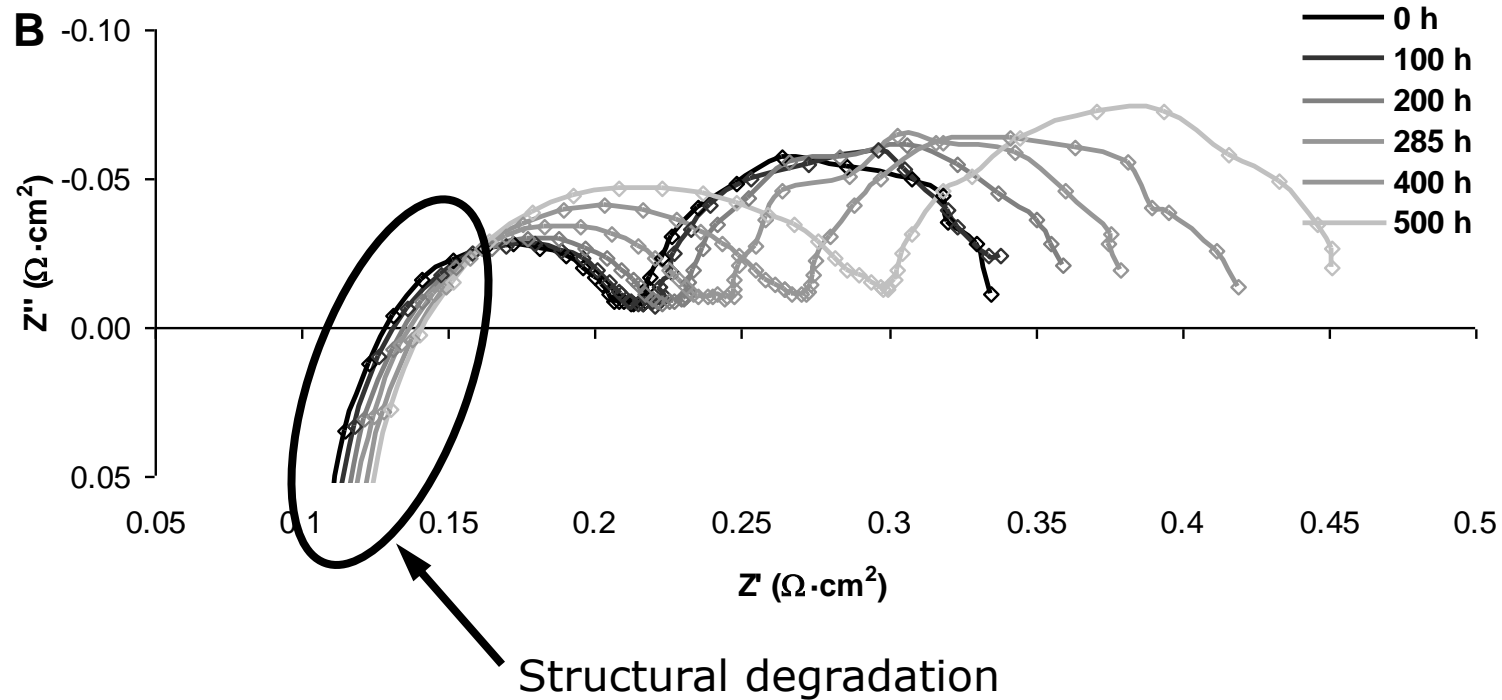
## — Today



Conditions: 850°C, -1,0 A/cm<sup>2</sup>, 45% CO<sub>2</sub> – 45% H<sub>2</sub>O – 10% H<sub>2</sub>

# Electrolysis durability at high current density

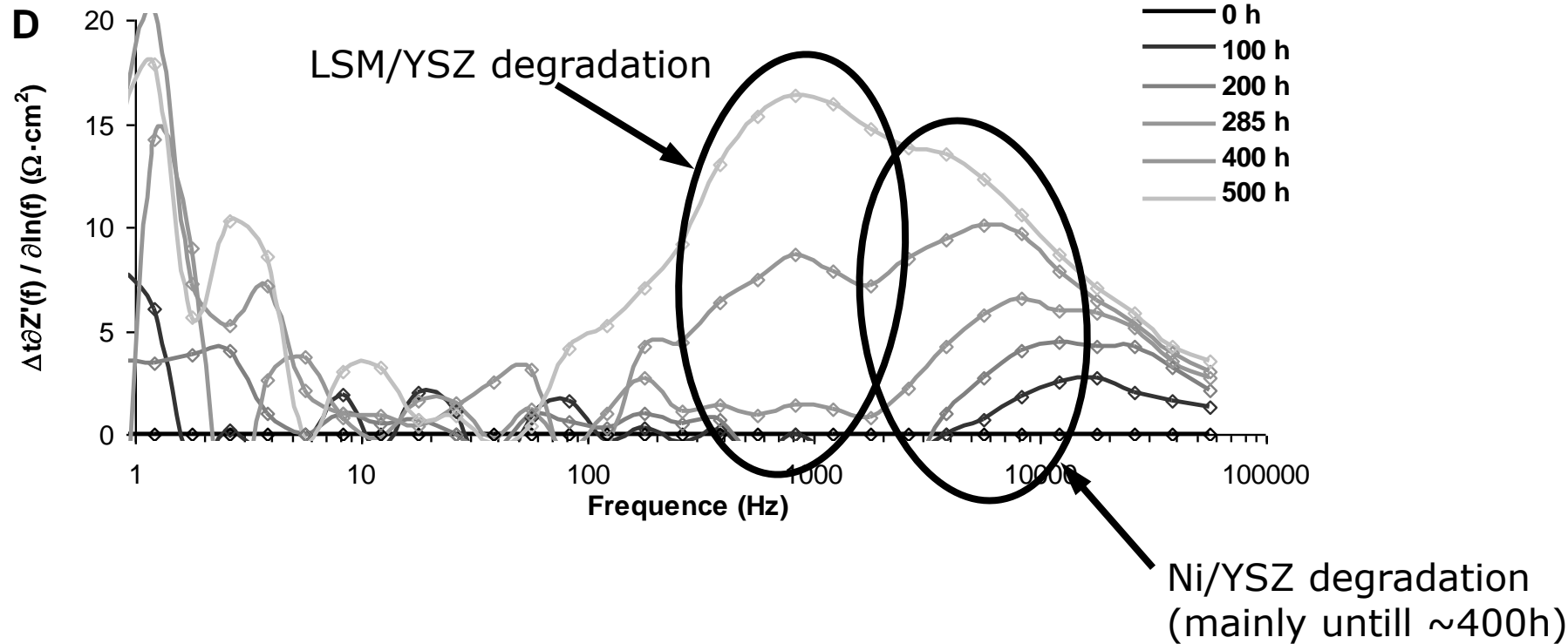
## — Today



Conditions: 850°C, -1,0 A/cm<sup>2</sup>, 45% CO<sub>2</sub> – 45% H<sub>2</sub>O – 10% H<sub>2</sub>

# Electrolysis durability at high current density

## — Today



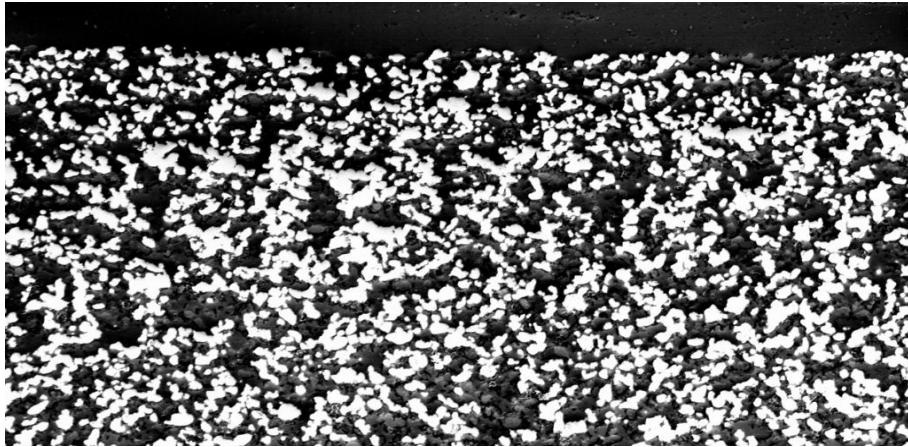
Conditions: 850°C, -1,0 A/cm<sup>2</sup>, 45% CO<sub>2</sub> – 45% H<sub>2</sub>O – 10% H<sub>2</sub>

# Electrolysis durability at high current density

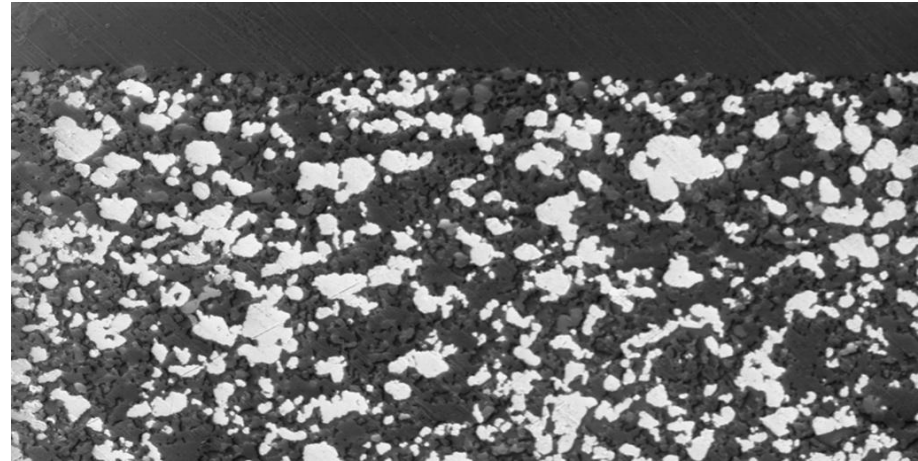
## — Today

Ni/YSZ electrode

Reference

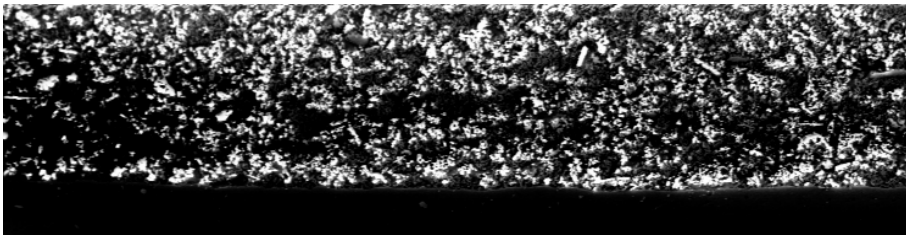


Tested cell

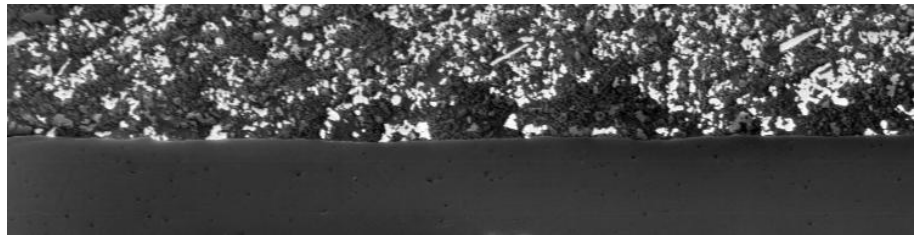


LSM/YSZ electrode

Reference



Tested cell



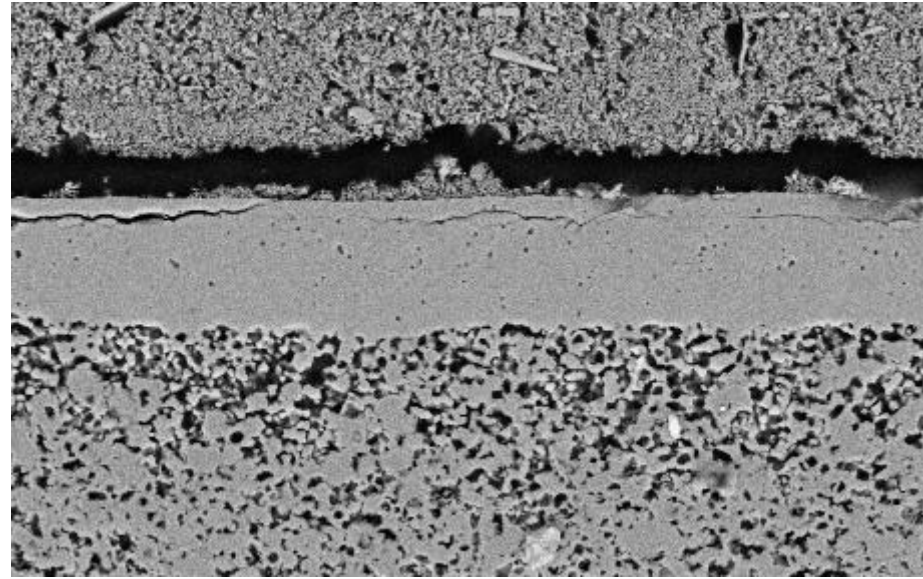
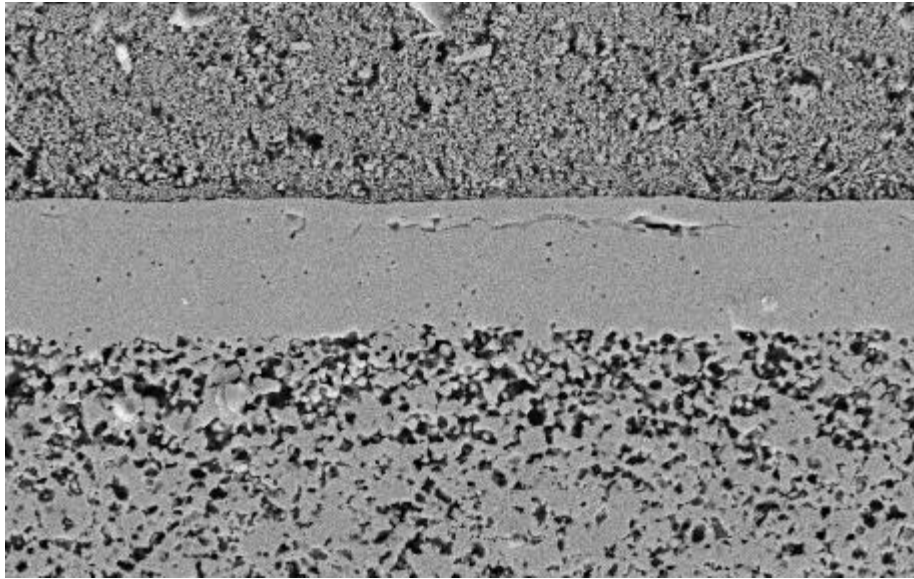
Conditions: 850°C, -1,0 A/cm<sup>2</sup>, 45% CO<sub>2</sub> – 45% H<sub>2</sub>O – 10% H<sub>2</sub>



# Electrolysis durability at high current density

## — Today

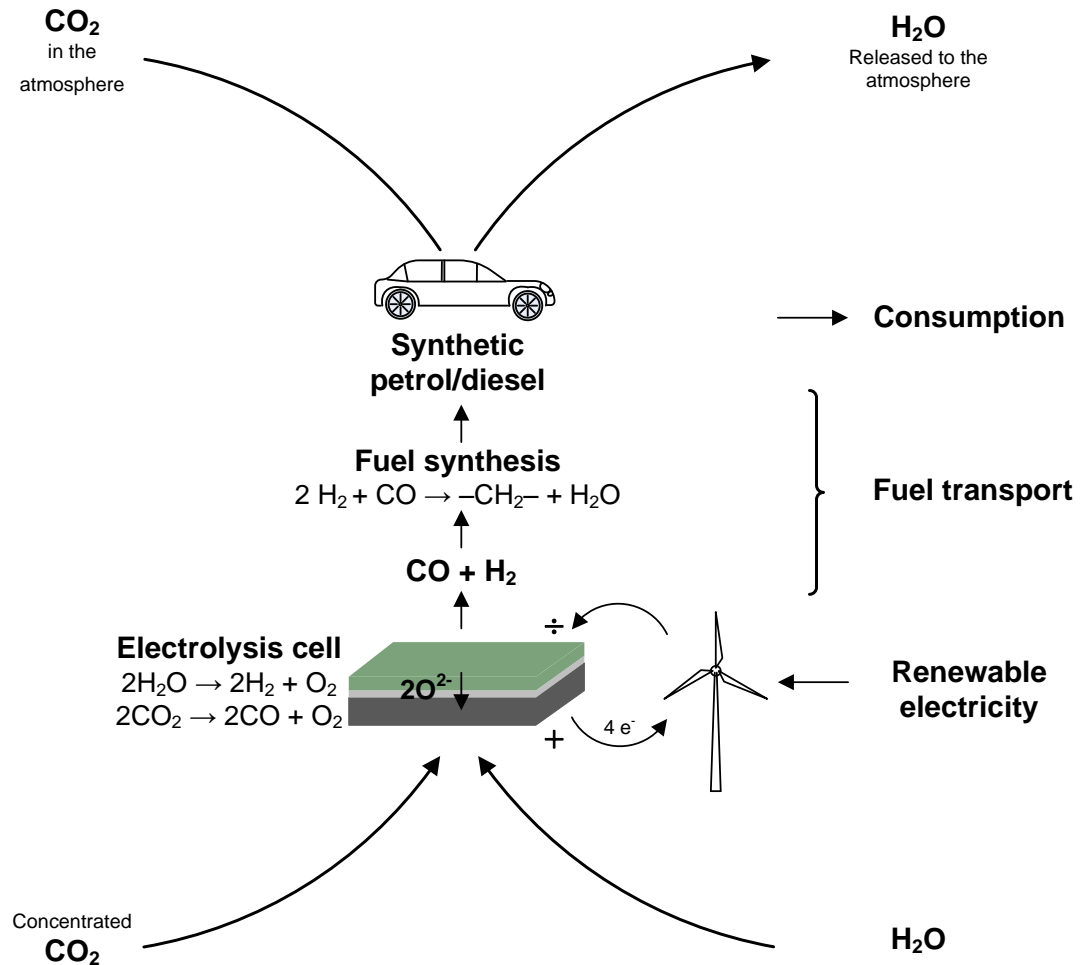
LSM/YSZ electrode



Conditions: 850°C, -1,0 A/cm<sup>2</sup>, 45% CO<sub>2</sub> – 45% H<sub>2</sub>O – 10% H<sub>2</sub>

# Vision

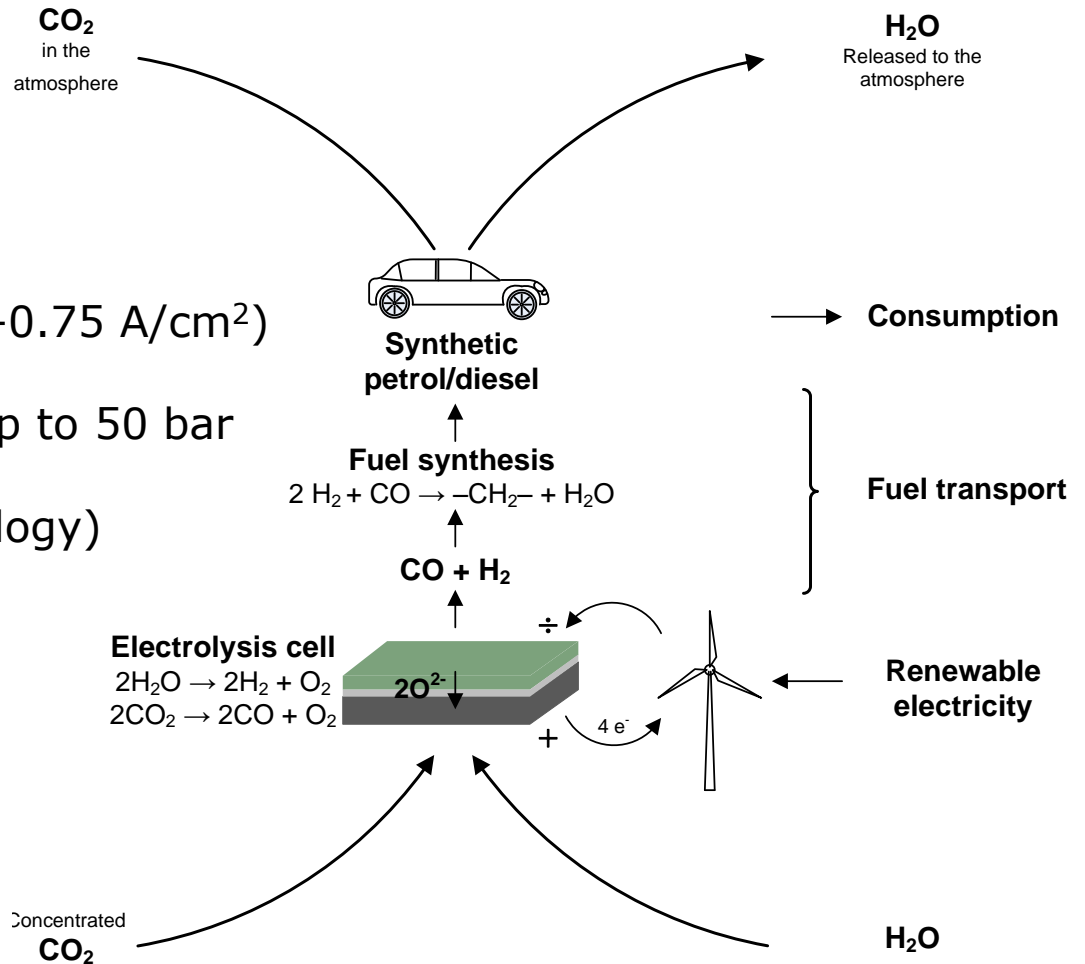
## — Synthesis at increased pressure



# Vision

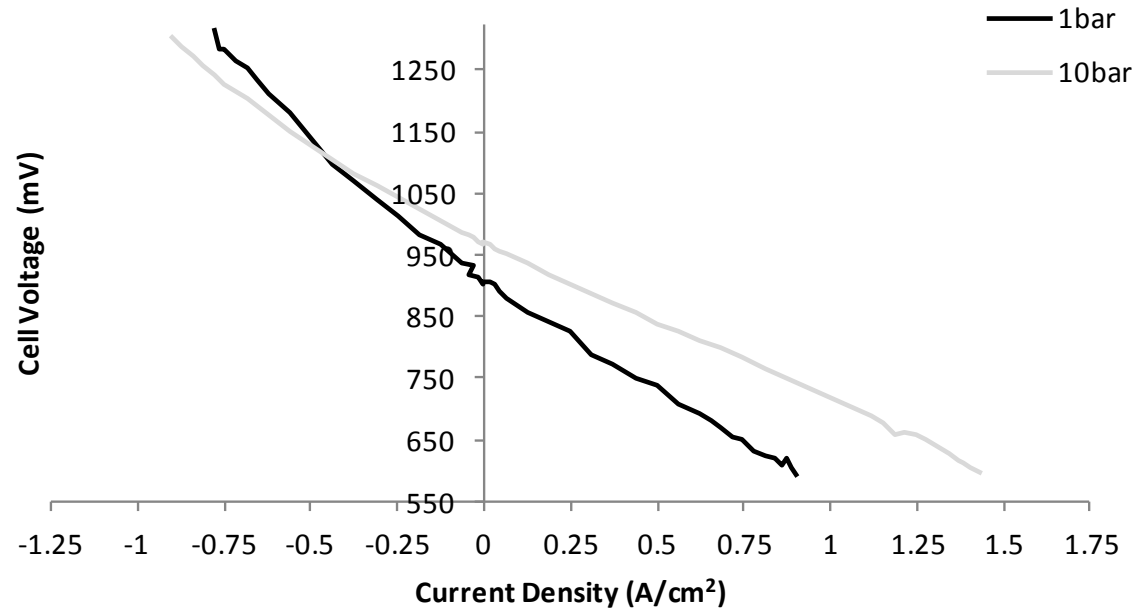
## — Synthesis at increased pressure

- 1) Durable stacks (proven up to  $-0.75 \text{ A/cm}^2$ )
- 2) Stacks operated at pressure up to 50 bar
- 3) Fuel synthesis (proven technology)



# Vision

— Synthesis at increased pressure



# Summary

- The Solid Oxide Cells are fully reversible

Fuel cell operation  $\longleftrightarrow$  Electrolysis operation

- Degradation is more severe in electrolysis mode compared to fuel cell mode
- Degradation at mild conditions is related to impurities

This degradation can be avoided by cleaning for impurities

- At harsh conditions structural changes occur in the cells

Need cells with lower polarisation  $\rightarrow$  lower degradation

- Cells can be operated safely at current densities up to  $-0.75 \text{ A/cm}^2$  at  $850^\circ\text{C}$
- Operation at high pressure advantageous for system integration
- Solid Oxide Electrolysis cells may contribute to storage of renewable electricity

# Acknowledgement

- The Danish National Advanced Technology Foundation's advanced technology platform  
"Development of 2nd generation bioethanol process and technology"
- Danish Council for Strategic Research, via the Strategic Electrochemistry Research Center
- European Commission via the project "Hi2H2"
- European Commission via the project "RelHy"
- EUDP via the project "Green Natural Gas"
- Danish PSO via the project "PlanSOEC"
- Danish PSO via the project "Durable solid oxide electrolysis cells and stacks"
- DTU Energy Conversion (Former Risø DTU) via the project SOECcell
- Topsoe Fuel Cell A/S (TOFC)